

Benzothiepin-derived molecular scaffolds for estrogen receptor modulators: synthesis and antagonistic effects in breast cancer cells

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(Received 8 March 2007; accepted 17 April 2007)

Abstract

A series of novel benzothiepin-derived compounds are described as potent selective modulators of the human estrogen receptor (SERMs). The objective of the study is to evaluate the antiproliferative effects of the compounds on human MCF-7 breast tumor cells. These heterocyclic compounds contain the traditional triarylethylene arrangement exemplified by tamoxifen, conformationally restrained through the incorporation of the benzothiepin ring system. The compounds demonstrated potency at nanomolar concentrations in antiproliferative assays against an MCF-7 human breast cancer cell line with low cytotoxicity. The compounds exhibited low nanomolar binding affinity for the estrogen receptor (ER) with some specificity for ER β , and also demonstrate potent antiestrogenic properties in the human uterine Ishikawa cell line. The effect of a number of functional group substitutions on the ER binding properties of the benzothiepin molecular scaffold is explored through a brief computational structure-activity relationship investigation with molecular simulation.

Keywords: Benzothiepin, estrogen receptor antagonists, antiproliferative activity, SERMs

Introduction

The estrogen receptor (comprising of two subtypes ER α and ER β) is a ligand inducible nuclear receptor which plays a critical physiological role as mediator of the actions of the estrogen hormones[1]. Tamoxifen **1** is a well established antagonist for the estrogen receptor and is the endocrine drug of choice in the treatment of ER positive breast cancer.[2–4] While the DNA binding domains of the two ER subtypes are well conserved, the amino acid sequence conservation in the ligand binding domain(LBD) is only 59%. The LBD volume for ER β is smaller than for ER α and there are important differences in the amino acids of the LBD including replacement of Met421 and Leu384 in ER α with Ile and Met in ER β [5,6]. Recent studies indicate that ER β expression may have a potential protective effect on normal cells against ER α induced hyperproliferation[7].

The determination of a number of crystal structures of ER-ligand complexes in recent years (e.g. 4-hydroxytamoxifen **1b** and raloxifene **2**)[8–12] together with the discovery of alternative novel scaffolds for the estrogen receptor modulators through “scaffold hopping” protocols[13] has facilitated the design of several novel agonist and antagonist-type ligands for the ER. Many examples of ER modulator scaffolds based on oxygen-containing core heterocyclic systems have been reported e.g. centchroman[14], benzopyrans such as EM-652 **3** which is the active metabolite of EM-800[15], benzopyranones, dibenzo[b,d]pyran-6-ones, bisbenzopyrans[16] (e.g. **4**), dihydrobenzoxathins[17], (e.g. **5**), bicyclo[3.3.1]nonene[18] and oxachrysenol[19] while mono[20] and bis-benzo[b]oxepines[21] have been designed as agonists of the estrogen receptor. The structures of estradiol together with selected SERMs are illustrated in Figure 1. We have recently identified a novel

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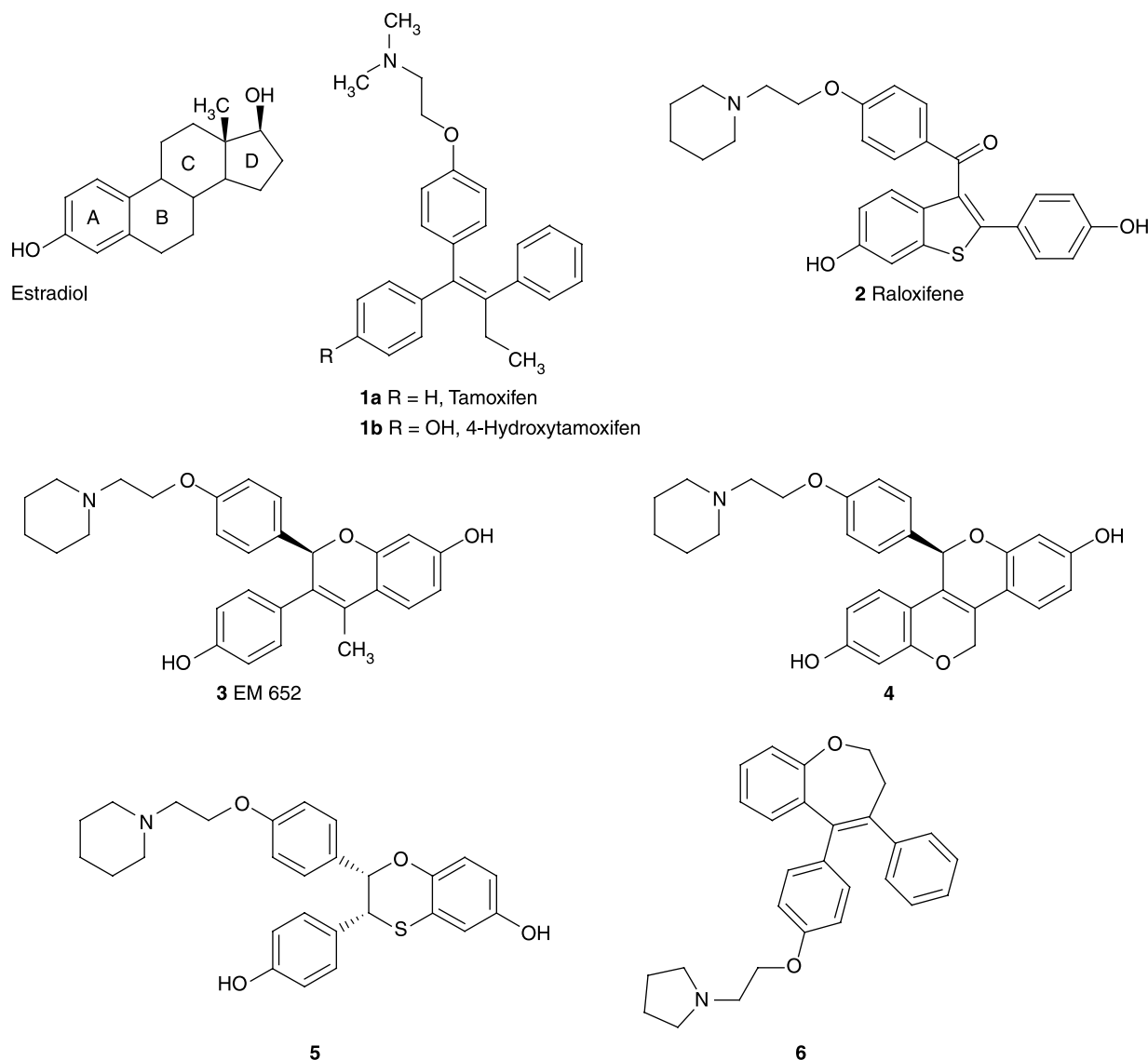


Figure 1. Structure of estradiol and SERMs.

estrogen receptor modulator core containing a central benzoxepin scaffold, (e.g. compound **6**), [20] which demonstrated some potential as antiproliferative and antagonist ER binding ligands and now report the further development of this novel core scaffold structure as tissue and subtype selective estrogen receptor modulators[22]. The synthesis of benzothiepin derivatives which are substituted at C-4, C-5 and C-8 positions is now described together with an evaluation of their antiproliferative activity and the relative binding affinities for ER α and ER β .

Materials and methods

Chemistry

IR spectra were recorded as thin films on NaCl plates on a Perkin-Elmer Paragon 100 FT-IR spectrometer.

^1H and ^{13}C NMR spectra were obtained on a Bruker Avance DPX 400 instrument at 20°C, 400.13 MHz for ^1H spectra, 100.61 MHz for ^{13}C spectra, in either CDCl_3 (internal standard tetramethylsilane) or CD_3OD . Low resolution mass spectra were run on a Hewlett-Packard 5973 MSD GC-MS system in an electron impact mode, while high resolution accurate mass determinations for all final target compounds were obtained on a Micromass Time of Flight mass spectrometer (TOF) equipped with electrospray ionization (ESI) interface operated in the positive ion mode at the High Resolution Mass Spectrometry Laboratory in the Department of Chemistry, Trinity College Dublin. Flash chromatography was carried out using standard silica gel 60 (230–400 mesh) obtained from Merck. All products isolated were homogenous on TLC. Compound **6** was prepared as we previously reported[20].

2,3-Dihydro-5-(4-hydroxyphenyl)-benzo[b]thiepin 9a. Under nitrogen atmosphere at -78°C , *n*-butyllithium (10 mL, 2.5 M in hexane) was added dropwise over 15 min to a solution of *p*-tetrahydropyranyloxy-bromobenzene **7** (25.0 mmol) in dry THF (40 mL) and stirred for 30 min. A solution of the 3,4-dihydro-1-(2HNMR)-benzo[b]thiepin-5-one **8a**[23] (25 mmol) in dry THF (40 mL) was added dropwise *via* a syringe. The reaction mixture was kept at this temperature for another 2 h and stirring continued overnight at room temperature. The mixture was partitioned between water (100 mL) and ethyl acetate (100 mL), the layers separated and the aqueous layer extracted with ethyl acetate (2 \times 100 mL) to afford the tertiary alcohol as a yellow oil. The crude product was taken up in MeOH (100 mL) and concentrated HCl (4 mL) and stirred at $60-70^{\circ}\text{C}$ for about 30 min, then partitioned between ethyl acetate and water (100 mL each), dried over MgSO_4 and concentrated *in vacuo*. Purification of the crude product was achieved by column chromatography (SiO_2 ; ethyl acetate/hexane 1:10, v/v). The product was obtained as a yellow oil (60%) which was used in subsequent reactions without further purification. IR ν_{max} (CHCl_3) cm^{-1} 3364 (w, br; O—H), 1610 (s; C=C); ^1H NMR (400 MHz, MeOD-d_4) δ 2.21 (q; 2H; $^3\text{J} = 6.5$ Hz; $-\text{CH}-\text{CH}_2-$), 3.43 (t; 2H; $^3\text{J} = 6.5$ Hz; $-\text{S}-\text{CH}_2-$), 6.46 (t; 1H; $^3\text{J} = 7.8$ Hz; $-\text{CH}-\text{CH}_2-$), 6.73/7.03 (AA'BB'; 4H; $^3\text{J} = 8.8$ Hz; Phenyl-H), 7.03 (1H merged; H9), 7.22 (dt; 1H; $^3\text{J} = 7.4$ Hz, $^4\text{J} = 1.8$ Hz; H7), 7.30 (dt; 1H; $^3\text{J} = 7.4$ Hz, $^4\text{J} = 1.4$ Hz; H8), 7.64 (dd; 1H; $^3\text{J} = 7.5$ Hz, $^4\text{J} = 1.5$ Hz; H6); ^{13}C NMR (MeOD-d_4 , 100 MHz) δ 26.0 (CH_2), 42.8 (CH_2), 114.1 (2 \times CH), 126.1 (CH), 126.7 (CH), 127.3 (CH), 128.3 (2 \times CH), 130.0 (CH), 133.0 (C), 134.0 (CH), 134.3 (C), 143.2 (C), 145.4 (C), 156.2 (C—O).

2,3-Dihydro-5-(4-hydroxyphenyl)-8-methoxybenzo[b]thiepin 9b. The procedure was carried out as above using 3,4-dihydro-8-methoxy-1-(2H)-benzo[b]thiepin-5-one **8b** and *p*-tetrahydropyranyloxy-bromobenzene. Purification was achieved by column chromatography (SiO_2 ; ethyl acetate / petroleum ether 40–60 1:10, v/v). The product was obtained as a yellow oil (60%) which was used in subsequent reactions without further purification. IR ν_{max} (film) cm^{-1} 3390 (w, br; O—H), 1609 (s; C=C). ^1H NMR (400 MHz, MeOD-d_4) δ 2.22 (q; 2H; $^3\text{J} = 6.9$ Hz; $-\text{CH}-\text{CH}_2-$), 3.42 (t; 2H; $^3\text{J} = 6.5$ Hz; $-\text{S}-\text{CH}_2-$), 3.82 (s; 3H; OCH_3), 6.38 (t; 1H; $^3\text{J} = 7.8$ Hz; $-\text{CH}-\text{CH}_2-$), 6.72/7.02 (AA'BB'; 4H; $^3\text{J} = 8.8$ Hz; Phenyl-H), 6.88 (dd; 1H; $^3\text{J} = 8.5$ Hz, $^4\text{J} = 2.5$ Hz; H7), 6.94 (d; 1H; $^3\text{J} = 8.5$ Hz; H6), 7.21 (d; 1H; $^4\text{J} = 2.5$ Hz; H9) ^{13}C NMR (MeOD-d_4 , 100 MHz) δ 26.1 (CH_2), 42.7 (CH_2), 54.0 (CH_3), 113.2 (CH), 114.0 (2 \times CH), 118.5 (CH), 125.2 (CH), 128.3 (2 \times CH), 131.0 (CH),

133.3 (C), 135.5 (C), 137.3 (C), 143.0 (C), 156.2 (C—O), 158.0 (C—O).

2,3-Dihydro-8-fluoro-5-(4-hydroxyphenyl)-benzo[b]thiepin 9c. Preparation as above from 8-fluoro-3,4-dihydro-2H-1-benzothiepin-5-one, **8c** (0.925 g, 4.72 mmol) The product was purified by chromatography (silica, 5% diethyl ether/hexane) to give the product as a white solid (0.564 g, 44%) which was used in subsequent reactions without further purification. ν (KBr): 3326 (OH), 1266 (CM_{-1}); ^1H (400 MHz, CDCl_3 , Me_4Si): δ 7.40 (1 H, dd, $J = 9.0, 2.5$), 7.10–6.97 (4 H, m, aromatic CH), 6.76 (2 H, d, $J = 8.5$, aromatic CH), 6.47 (1 H, t, $J = 7.5$, CH), 3.48 (2 H, t, $J = 7.0$, OCH_2), 2.25 (2 H, t, $J = 7.0$, CH_2), ppm; ^{13}C NMR (101 MHz, CDCl_3 , Me_4Si): δ 155.1 (C), 142.7 (C), 134.6 (C), 132.2 (C), 132.1 (CH), 129.3 (CH), 127.5 (CH), 121.4 (CH), 121.2 (CH), 115.3 (CH), 115.1 (CH), 44.1 (CH_2), 26.9 (CH_2) ppm; ^{19}F (376 MHz, CDCl_3 , Me_4Si): -114.7 ppm.

2,3-Dihydro-5-[4-(2-pyrrolidinylethoxy)-phenyl]-benzo[b]thiepin 11a. A mixture of 2,3-dihydro-5-(4-hydroxyphenyl)-benzo[b]thiepin **9a** (3.52 mmol) and potassium carbonate (17.60 mmol) in acetone (40 mL) were heated for 1 h. Then, 1-(2-chloroethyl)-pyrrolidine hydrochloride (7.03 mmol) was added in portions and the mixture refluxed for 24 h. Finally the slightly brown solution was cooled to room temperature and filtered. The solid remainders were washed thoroughly with acetone and the solvent of the combined filtrates removed *in vacuo*. The crude product was purified by column chromatography (SiO_2 ; methanol/dichloromethane 1:25, v/v) to afford the product as a yellow oil (66%) which was used in subsequent reactions without further purification. ^1H NMR (400 MHz, MeOD-d_4) δ 1.84–1.87 (m; 4H; $-(\text{CH}_2)_2-\text{CH}_2-\text{N}-$), 2.23 (q; 2H; $^3\text{J} = 7.0$ Hz; $-\text{CH}-\text{CH}_2-$), 2.70 (m; 4H; $\text{CH}_2-\text{N}-\text{CH}_2-$), 2.95 (t; 2H; $^3\text{J} = 5.8$ Hz; $-\text{N}-\text{CH}_2-$), 3.44 (t; 2H; $^3\text{J} = 6.5$ Hz; $-\text{S}-\text{CH}_2-$), 4.14 (t; 2H; $^3\text{J} = 5.5$ Hz; $-\text{O}-\text{CH}_2-$), 6.50 (t; 1H; $^3\text{J} = 7.8$ Hz; $-\text{CH}-\text{CH}_2-$), 6.89/7.13 (AA'BB'; 4H; $^3\text{J} = 8.8$ Hz; Phenyl-H), 7.01 (dd; 1H; $^3\text{J} = 7.5$ Hz, $^4\text{J} = 1.5$ Hz; H9), 7.24 (dt; 1H; $^3\text{J} = 7.5$ Hz, $^4\text{J} = 1.5$ Hz; H7), 7.31 (dt; 1H; $^3\text{J} = 7.5$ Hz, $^4\text{J} = 1.5$ Hz; H8), 7.65 (d; 1H; $^3\text{J} = 7.5$ Hz; H6). ^{13}C NMR (MeOD-d_4 , 100 MHz) δ 22.3 (2 \times CH_2), 26.0 (CH_2), 42.7 (CH_2), 53.7 (2 \times CH_2-N), 54.0 (CH_3), 54.2 (CH_2-N), 65.7 (CH_2-O), 113.4 (2 \times CH), 126.8 (CH), 126.8 (CH), 127.3 (CH), 128.3 (2 \times CH), 130.0 (CH), 134.0 (CH), 134.4 (C), 142.9 (C), 145.2 (C), 157.8 (C—O)(M^+) $\text{C}_{22}\text{H}_{25}\text{NOS}$ (351.51)

2,3-Dihydro-8-methoxy-5-[4-(2-pyrrolidinylethoxy)-phenyl]-benzo[b]thiepin 11b. Preparation as above from

2,3-dihydro-5-(4-hydroxyphenyl)-8-methoxy-benzo[b]thiepin **9b**. The crude product was purified by column chromatography (SiO₂; methanol/dichloromethane 1:25, v/v) to afford the product as a yellow oil (74%) which was used in subsequent reactions without further purification. ¹H NMR (400 MHz, MeOD-d₄) δ 1.84–1.88 (m; 4H; $-(CH_2)_2-CH_2-N-$), 2.25 (q; 2H; ³J = 7.2 Hz; $-CH-CH_2-$), 2.72 (m; 4H; CH_2-N-CH_2-), 2.97 (t; 2H; ³J = 5.5 Hz; $-N-CH_2-$), 3.44 (t; 2H; ³J = 7.0 Hz; $-S-CH_2-$), 3.85 (s; 3H; OCH₃), 4.14 (t; 2H; ³J = 5.8 Hz; $-O-CH_2-$), 6.42 (t; 1H; ³J = 7.8 Hz; $-CH-CH_2-$), 6.88 (1H merged; H7), 6.88/7.13 (AA'BB'; 4H; ³J = 8.8 Hz; Phenyl-H), 6.93 (d; 1H; ³J = 8.5 Hz, H6), 7.22 (d; 1H; ⁴J = 2.5 Hz, H9) ¹³C NMR (MeOD-d₄, 100 MHz) δ 22.3 (2 × CH₂), 26.2 (CH₂), 42.6 (CH₂), 53.7 (2 × CH₂-N), 54.0 (CH₃), 54.1 (CH₂-N), 65.6 (CH₂-O), 113.2 (CH), 113.3 (2 × CH), 118.5 (CH), 126.0 (CH), 128.3 (2 × CH), 130.9 (CH), 134.7 (C), 135.6 (C), 137.1 (C), 142.7 (C), 157.7 (C-O), 158.1 (C-O).

4-Bromo-2,3-Dihydro-5-[4-(2-pyrrolidinylethoxy)-phenyl]-benzo[b]thiepin 12a. To a solution of 2,3-dihydro-5-[4-(2-pyrrolidinylethoxy)-phenyl]-benzo[b]thiepin **11a** (2.11 mmol) in dry dichloromethane (30 mL) at -10°C was added in small portions pyridiniumbromide perbromide (2.11 mmol) at such a rate, that the reagent was readily dissolved before the next addition. The reaction mixture was then stirred at room temperature for 8 h. A solution of sodium hydrogen carbonate (10%, 50 mL) was added and the aqueous layer extracted with dichloromethane (3 × 50 mL). The combined organic layers were washed with water and brine (50 mL each), dried over MgSO₄ and the solvent removed under reduced pressure. The crude yellow liquid (containing one equivalent of pyridine) was purified by column chromatography (SiO₂; methanol/dichloromethane 1:25, v/v) to afford the product as a yellow oil (82%), which was used in subsequent reactions without further purification. IR ν_{max} (film) cm⁻¹ 2787 (s), 1606 (s; C=C); ¹H NMR (400 MHz, MeOD-d₄) δ 1.84–1.88 (m; 4H; $-(CH_2)_2-CH_2-N-$), 2.72 (m; 4H; CH_2-N-CH_2-), 2.80 (t; 2H; ³J = 6.5 Hz; $-CH_2-CH_2-S-$), 2.97 (t; 2H; ³J = 5.5 Hz; $-N-CH_2-$), 3.59 (t; 2H; ³J = 6.5 Hz; $-S-CH_2-$), 4.15 (t; 2H; ³J = 5.5 Hz; $-O-CH_2-$), 6.92/7.13 (AA'BB'; 4H; ³J = 8.8 Hz; Phenyl-H), 6.90–6.92 (m; 1H; H9), 7.20–7.27 (m; 2H; H7, H8), 7.61–7.64 (m; 1H; H6); ¹³C NMR (MeOD-d₄, 100 MHz) δ 22.3 (2 × CH₂), 37.6 (CH₂), 40.5 (CH₂), 53.7 (2 × CH₂-N), 54.1 (CH₂-N), 65.7 (CH₂-O), 113.1 (2 × CH), 122.4 (C-Br), 127.3 (CH), 127.3 (CH), 130.0 (CH), 130.3 (2 × CH), 132.8 (C), 134.1 (C), 134.2 (CH), 141.2 (C), 146.0 (C), 157.7 (C-O).

4-Bromo-2,3-dihydro-8-methoxy-5-[4-(2-pyrrolidinylethoxy)-phenyl]-benzo[b]thiepin 12b. Preparation from 2,3-dihydro-8-methoxy-5-[4-(2-pyrrolidinylethoxy)-phenyl]-benzo[b]thiepin **11b** as described above. The crude material was purified by column chromatography (SiO₂; methanol/dichloromethane 1:25, v/v) to afford the product as a yellow oil (74%) which was used in subsequent reactions without further purification. IR ν_{max} (film) cm⁻¹ 2784 (s), 1606 (s; C=C); ¹H NMR (400 MHz, MeOD-d₄) δ 1.85–1.88 (m; 4H; $-(CH_2)_2-CH_2-N-$), 2.73 (m; 4H; CH_2-N-CH_2-), 2.82 (t; 2H; ³J = 6.5 Hz; $-CH_2-CH_2-S-$), 2.97 (t; 2H; ³J = 5.5 Hz; $-N-CH_2-$), 3.58 (t; 2H; ³J = 6.5 Hz; $-S-CH_2-$), 3.80 (s; 3H; OCH₃), 4.16 (t; 2H; ³J = 5.5 Hz; $-O-CH_2-$), 6.80–6.82 (m; 2H; H6, H7), 6.91/7.12 (AA'BB'; 4H; ³J = 9.0 Hz; Phenyl-H), 7.18–7.19 (m; 1H; H9). ¹³C NMR (MeOD-d₄, 100 MHz) δ 22.3 (2 × CH₂), 37.7 (CH₂), 40.2 (CH₂), 53.7 (2 × CH₂-N), 54.0 (CH₃), 54.1 (CH₂-N), 65.6 (CH₂-O), 113.0 (2 × CH), 113.4 (CH), 118.7 (CH), 121.2 (C-Br), 130.3 (2 × CH), 131.1 (CH), 134.1 (C), 134.3 (C), 137.9 (C), 140.9 (C), 157.6 (C-O), 158.4 (C-O).

4-Bromo-8-fluoro-5-[4-(2-pyrrolidinylethoxy)-phenyl]-2,3-dihydrobenzo[b]thiepin 10. Preparation from **9c** (0.360 g, 1.32 mmol) as described above for the synthesis of compounds **12a–b**. The solvent was removed under reduced pressure to give the bromide as a yellow oil which required no purification (91%). ν_{max} (film): 3355 (OH), 1258 NMR, 731 NMR; ¹H (400 MHz, CDCl₃, Me₄Si): δ 7.08 (2 H, d, \mathcal{J} = 8.5), 6.84–6.63 (5 H, m, aromatic CH), 4.58 (2 H, t, \mathcal{J} = 6.0, OCH₂CH₂), 3.00 (2 H, t, \mathcal{J} = 6.0, OCH₂CH₂) ppm; ¹³C NMR (101 MHz, CDCl₃, Me₄Si): δ 155.1 (C), 142.7 (C), 134.6 (C), 132.2 (C), 132.1 (CH), 129.3 (CH), 127.5 (CH), 121.4 (CH), 121.2 (CH), 115.3 (CH), 44.1 (CH₂), 26.9 (CH₂) ppm; ¹⁹F (376 MHz, CDCl₃, Me₄Si): -113.9 ppm.

2,3-Dihydro-4-(4-hydroxyphenyl)-5-[4-(2-pyrrolidinylethoxy)-phenyl]-benzo[b]thiepin 13a. Pd(PPh₃)₄ (0.035 mmol) was added to a solution of 4-bromo-2,3-dihydro-5-[4-(2-pyrrolidinylethoxy)-phenyl]-benzo[b]thiepin **12a**. (1.16 mmol), 4-hydroxyphenylboronic acid (1.74 mmol), and 2 M Na₂CO₃ (5.80 mmol) in THF (20 mL) and heated to reflux for 5–6 h. After cooling, the mixture was partitioned between water and ethyl acetate (40 mL each), filtered and extracted with ethyl acetate (3 × 40 mL). The combined organic layers were washed with water and brine (40 mL each), dried over MgSO₄ and the solvent removed *in vacuo*. The orange product was purified by column chromatography (SiO₂; methanol/dichloromethane 1:15, v/v) to afford the product which was recrystallised from MeOH (and

few drops of DCM) at 4°C as white solid; (72%), M.p. 178–180°C (dec.); IR ν_{\max} (KBr) cm^{-1} (3491w, br; –OH), 1608(s; C=C)1508,1239. ^1H NMR (400 MHz, DMSO- d_6) δ 1.67–1.70 (m; 4H; $-(\text{CH}_2)_2-\text{CH}_2-\text{N}-$), 2.54–2.57 (m; 6H; $-\text{CH}_2-\text{CH}_2-\text{S}-$, $-\text{CH}_2-\text{N}-\text{CH}_2-$), 2.81 (t; 2H; $^3\text{J} = 5.6 \text{ Hz}$; $-\text{N}-\text{CH}_2-$), 3.35 (t; 2H; $^3\text{J} = 6.0 \text{ Hz}$; $-\text{S}-\text{CH}_2-$), 3.99 (t; 2H; $^3\text{J} = 5.5 \text{ Hz}$; $-\text{O}-\text{CH}_2-$), 6.59 (d; 2H; $^3\text{J} = 8.5 \text{ Hz}$; Phenyl-H), 6.68–6.72 (m; 4H; Phenyl-H), 7.00 (d; 2H; $^3\text{J} = 8.5 \text{ Hz}$; Phenyl-H), 6.85 (dd; 1H; $^3\text{J} = 8.5 \text{ Hz}$, $^4\text{J} = 1.5 \text{ Hz}$; H9), 7.23 (dt; 1H; $^3\text{J} = 7.5 \text{ Hz}$, $^4\text{J} = 1.8 \text{ Hz}$; H7), 7.30 (dt; 1H; $^3\text{J} = 7.5 \text{ Hz}$, $^4\text{J} = 1.3 \text{ Hz}$; H8), 7.61 (dd; 1H; $^3\text{J} = 7.5 \text{ Hz}$, $^4\text{J} = 1.5 \text{ Hz}$; H6), 9.39 (s; 1H; –OH). ^{13}C NMR (DMSO- d_6 , 100 MHz) δ 23.0 (2 \times CH_2), 34.3 (CH_2), 42.1 (CH_2), 54.0 (2 \times CH_2-N), 54.9 (CH_2-N), 66.0 (CH_2-O), 113.7 (2 \times CH), 114.9 (2 \times CH), 127.5 (CH), 128.3 (CH), 130.4 (2 \times CH), 130.9 (CH), 131.8 (2 \times CH), 132.0 (C), 133.2 (C), 134.4 (CH), 134.5 (C), 136.9 (C), 140.4 (C), 148.0 (C), 156.1 (C–O), 156.7 (C–O). $\text{C}_{28}\text{H}_{29}\text{NO}_2\text{S}$. 0.3 CH_2Cl_2 Calculated: C: 72.46H: 6.36, N: 2.99; Found: C: 72.15, H: 6.40, N: 2.86. HRMS: Found: 444.1999 ($\text{M}^+ + 1$); $\text{C}_{28}\text{H}_{30}\text{NO}_2\text{S}$ requires 444.1997.

2,3-Dihydro-4-hydroxyphenyl-8-methoxy-5-[4-(2-pyrrolidinylethoxy)-phenyl]-benzo [b] thiepin 13b. Preparation as above from 4-bromo-2,3-dihydro-8-methoxy-5-[4-(2-pyrrolidinylethoxy)-phenyl]benzo [b]thiepin **12b**. The dark red crude product was purified by column chromatography (SiO_2 ; methanol / dichloromethane 1:15, v/v) to afford the product as a beige solid, which was dissolved in DCM, treated with 6M HCl and set aside at 4°C to achieve crystallisation of product as the hydrochloride salt. The product was obtained as a white solid; (58%), M.p. 246–249°C (dec.); IR ν_{\max} (KBr) cm^{-1} 3743(w, br; –OH)1608, (s; C=C)1508,1239. ^1H NMR (400 MHz, DMSO- d_6) δ 1.92 (br; 4H; $-(\text{CH}_2)_2-\text{CH}_2-\text{N}-$), 2.55 (t; 2H; $^3\text{J} = 6.5 \text{ Hz}$; $-\text{CH}_2-\text{CH}_2-\text{S}-$), 3.09 (br; 4H; $-\text{CH}_2-\text{N}-\text{CH}_2-$), 3.50 (m; 4H; $-\text{N}-\text{CH}_2-$, $-\text{S}-\text{CH}_2-$), 3.79 (s; 3H; $-\text{O}-\text{CH}_3$), 4.23 (t; 2H; $^3\text{J} = 4.8 \text{ Hz}$; $-\text{O}-\text{CH}_2-$), 6.59 (d; 2H; $^3\text{J} = 8.5 \text{ Hz}$; Phenyl-H), 6.74–6.76 (m; 5H; Phenyl-H), 6.89 (dd; 1H; $^3\text{J} = 8.5 \text{ Hz}$, $^4\text{J} = 2.5 \text{ Hz}$; H7), 6.99 (d; 2H; $^3\text{J} = 8.0 \text{ Hz}$; Phenyl-H), 7.17 (d; 1H; $^4\text{J} = 2.5 \text{ Hz}$; H9), 9.40 (s; 1H; –OH) ^{13}C NMR (DMSO- d_6 , 100 MHz) δ 23.0 (2 \times CH_2), 34.4 (CH_2), 41.9 (CH_2), 54.0 (2 \times CH_2-N), 54.9 (CH_2-N), 55.2 (CH_3), 66.1 (CH_2-O), 113.6 (2 \times CH), 114.4 (CH), 114.9 (2 \times CH), 118.7 (CH), 130.4 (2 \times CH), 131.8 (2 \times CH), 131.9 (CH), 132.2 (C), 134.4 (C), 134.6 (C), 136.7 (C), 139.4 (C), 140.0 (C), 156.0 (C–O), 156.6 (C–O), 157.7 (C–O). $\text{C}_{29}\text{H}_{31}\text{NO}_3\text{S}$. 0.3 CH_2Cl_2 ; Calculated: C: 65.70, H: 6.13, N: 2.63; Found: C: 65.76; H: 6.20, N: 2.58. HRMS: Found: 474.2107, ($\text{M}^+ + 1$), $\text{C}_{29}\text{H}_{32}\text{NO}_3\text{S}$ requires 474.2103.

1-{2-[4-(4-Bromo-8-fluoro-2,3-dihydro-1-benzo[b]thiepin-5-yl)-phenoxy]-ethyl}-pyrrolidine 12c. 4-(4-Bromo-8-fluoro-2,3-dihydro-1-benzothiepin-5-yl)-phenol **10** (0.46 g, 1.32 mmol) was treated with potassium carbonate (0.91 g, 6.60 mmol) and 1-(2-chloroethyl)pyrrolidine hydrochloride (0.34 g, 1.98 mmol) in acetone (20 ml) as above. The crude product was purified by chromatography (2.5% methanol/dichloromethane) to give the product as a yellow oil (0.236 g, 40%). IR ν_{\max} (film) cm^{-1} : 2960, 1606 (C = C), 1508, 1472, 1246, 732; ^1H (400 MHz, CDCl_3 , Me_4Si): δ 7.35 (1 H, dd, $\text{J} = 8.0, 1.5$, C_6H_4), 7.14 (2 H, d, $\text{J} = 8.5$, C_6H_4), 6.86–6.91 (4 H, m, C_6H_4), 4.11 (2 H, t, avg. $\text{J} = 6.0$, $\text{OCH}_2\text{CH}_2\text{N}-$), 3.62, (2 H, t, $\text{J} = 6.5$, OCH_2CH_2), 2.91 (2 H, t, $\text{J} = 6.0$, $\text{OCH}_2\text{CH}_2\text{N}-$), 2.82 (2 H, $\text{J} = 6.5$, OCH_2CH_2), 2.62 (4 H, m, CH_2NCH_2), 1.81, (4 H, m, CH_2CH_2); ^{13}C (101 MHz, CDCl_3 , Me_4Si): δ 162.2 (C), 159.7 (C), 158.2 (C), 140.7 (C), 134.1 (C), 132.4 (CH), 130.9 (CH), 123.0 (C), 121.5 (CH), 121.3 (CH), 115.5 (CH), 115.3 (CH), 113.9 (CH), 66.8 (CH_2), 54.9 (CH_2), 41.6 (CH_2), 38.4 (CH_2), 23.4 (CH_2) ppm; ^{19}F (376 MHz, CDCl_3 , Me_4Si): $-\text{113.7}$ ppm; HRMS Found 448.0760; $\text{C}_{22}\text{H}_{24}\text{BrFNOS}$ requires 448.0746.

4-{8-Fluoro-5-[4-(2-pyrrolidin-1-yl-ethoxy)-phenyl]-2,3-dihydro-1-benzo[b]thiepin-4-yl}-phenol 13c. $\text{Pd}(\text{PPh}_3)_4$ (48 mg, 31 mmol) was added to a solution of 1-{2-[4-(4-bromo-8-fluoro-2,3-dihydro-1-benzo[b]thiepin-5-yl)-phenoxy]-ethyl}-pyrrolidine **12c** (138 mg, 310 μmol) in dry THF under nitrogen and stirred for 15 min. 4-Hydroxyphenylboronic acid (56 mg, 0.4 mmol) and 2 M sodium carbonate (0.77 ml, 1.55 mmol) was added and the solution refluxed for 6 h. The solution was cooled and acidified with 2 M hydrochloric acid. The aqueous layer was extracted with dichloromethane and the combined organic layers were washed with water (30 mL), brine (30 mL), dried over sodium sulfate and the solvent removed under reduced pressure. The product was purified by column chromatography (silica gel, 2.5% methanol/dichloromethane) to give the product as a brown solid (23 mg, 16%). ν_{\max} (KBr) cm^{-1} 3437 (OH), 1608 (C = C), 1508, 1247 (C=C); ^1H (400 MHz, d -DMSO, Me_4Si): δ 6.96 (2 H, d, $\text{J} = 8.5$, C_6H_3), 6.86–6.67 (7 H, m, aromatic CH), 6.56 (2 H, d, $\text{J} = 9.0$), 4.60 (2 H, t, avg. $\text{J} = 6.2$, $\text{OCH}_2\text{CH}_2\text{N}-$), 4.20 (2 H, t, $\text{J} = \text{avg. } 5.0$, OCH_2CH_2), 3.22 (6 H, m, 4 \times CH_2NCH_2 , 2 \times CH_2NCH_2), 2.68 (4 H, m, CH_2-CH_2); ^{13}C (101 MHz, CDCl_3 , Me_4Si): δ 155.9 (C), 144.3 (C), 140.5 (C), 136.3 (C), 135.7 (C), 135.6 (C), 134.1 (C), 132.2 (C), 131.6 (CH), 130.2 (CH), 129.9 (CH), 120.2, 119.9, 114.8, 144.6 (C–F), 114.4 (CH), 113.2 (CH), 65.6 (CH_2), 53.8 (CH_2), 53.7 (CH_2) 41.9 (CH_2), 34.0 (CH_2), 22.7 (CH_2); ^{19}F (376 MHz, CDCl_3 , Me_4Si): $-\text{117.17}$; Found 462.1914 $\text{C}_{28}\text{H}_{29}\text{NO}_2\text{F}$ S requires 462.1903.

2,3-Dihydro-8-hydroxy-4-hydroxyphenyl-5-[4-(2-pyrrolidinylethoxy)-phenyl]-benzo[b]thiepin **14**. A solution of 2,3-dihydro-8-methoxy-4-hydroxyphenyl-5-[4-(2-pyrrolidinylethoxy)-phenyl]-benzo[b]thiepin **13b** (52.8 μmol) in dry DCM (5 mL) at 0°C, was added dropwise to BBr_3 (0.26 mmol, 1 M solution in DCM) diluted with dry DCM (3 mL). After 10 h of stirring at 0°C the mixture was quenched with 10% (w/w) NaHCO_3 (20 mL) and the dark precipitate dissolved in a mixture of EtOAc/MeOH 10:1 (v/v) (20 mL) with vigorous stirring. The aqueous phase was separated and extracted with this EtOAc/MeOH-mixture. The combined organic fractions were dried with MgSO_4 and concentrated *in vacuo*. The crude product was purified by column chromatography (SiO_2 ; methanol/dichloromethane 1:7, v/v) to separate the product as a solid; (62%). $^1\text{H NMR}$ (400 MHz, DMSO-d_6) δ 1.73 (s, br; 4H; $-(\text{CH}_2)_2-\text{CH}_2-\text{N}-$), 2.53 (t; 2H; $^3\text{J} = 6.0\text{ Hz}$; $-\text{CH}_2-\text{CH}_2-\text{S}-$), 2.71 (s, br; 4H; $-\text{CH}_2-\text{N}-\text{CH}_2-$), 2.95 (s, br; 2H; $-\text{CH}_2-\text{N}-$), 3.31 (t; 2H; $^3\text{J} = 6.0\text{ Hz}$; $-\text{S}-\text{CH}_2-$), 4.03 (t; 2H; $^3\text{J} = 5.1\text{ Hz}$; $-\text{O}-\text{CH}_2-$), 6.57 (d; 2H; $^3\text{J} = 8.5\text{ Hz}$; Phenyl-H), 6.67–6.73 (m; 5H; Phenyl-H), 6.64 (d; 1H; $^3\text{J} = 8.5\text{ Hz}$, H6), 6.97 (d; 2H; $^3\text{J} = 8.5\text{ Hz}$; Phenyl-H), 7.03 (d; 1H; $^4\text{J} = 2.5\text{ Hz}$; H9), 9.34 (s; 1H; $-\text{OH}$), 9.70 (s; 1H; $-\text{OH}$) $^{13}\text{C NMR}$ (DMSO-d_6 , 100 MHz) δ 22.9 ($2 \times \text{CH}_2$), 34.5 (CH_2), 41.6 (CH_2), 53.9 ($2 \times \text{CH}_2-\text{N}$), 54.8 (CH_2-N), 65.5 (CH_2-O), 113.5 ($2 \times \text{CH}$), 114.9 ($2 \times \text{CH}$), 115.5 (CH), 120.4 (CH), 130.4 ($2 \times \text{CH}$), 131.8 ($2 \times \text{CH}$), 131.9 (CH), 132.4 (C), 134.1 (C), 135.0 (C), 136.9 (C), 138.2 (C), 138.9 (Aromatic C), 155.9 (C-O), 156.1 (C-O), 156.4 (C-O). HRMS Found 460.1959 ($\text{M}^+ + 1$); $\text{C}_{28}\text{H}_{30}\text{NO}_3\text{S}$ requires 460.1946.

Biochemical evaluation of activity:

Antiproliferation studies. All assays were performed in triplicate for the determination of mean values reported. Compounds were assayed as the free bases isolated from reaction. The human breast tumor cell line MCF-7 was cultured in Eagles minimum essential medium in a 95% O_2 /5% CO_2 atmosphere with 10% fetal calf serum. The medium was supplemented with 1% non-essential amino acids. Cells were trypsinised and seeded at a density of 1.5×10^4 into a 96-well plate and incubated at 37°C, 95% O_2 /5% CO_2 atmosphere for 24 h. After this time they were treated with 2 μL volumes of test compound which had been pre-prepared as stock solutions in ethanol to furnish the final concentration range of study, 1 nM–100 μM , and re-incubated for a further 72 h. Control wells contained the equivalent volume of the vehicle ethanol (1% v/v). The culture medium was then removed and the cells washed with 100 μL phosphate buffered saline (PBS) and 50 μL MTT added, to reach a final concentration of 1 mg/mL MTT added. Cells were incubated for 2 h in darkness at 37°C. At this point solubilization was begun through the

addition of 200 μL DMSO and the cells maintained at room temperature in darkness for 20 min to ensure thorough colour diffusion before reading the absorbance. The absorbance value of control cells (vehicle treated) was set to 100% cell viability and from this graphs of absorbance versus cell density per well were prepared to assess cell viability and from these, graphs of percentage cell viability versus concentration of subject compound added were drawn.

Cytotoxicity studies. Human MCF-7 breast cancer cells were plated at a density of 1.5×10^4 per well in a 96-well plate, then incubated at 37°C, 95% O_2 /5% CO_2 atmosphere for 24 h. Cells were treated with the compound of choice at varying concentrations (1 nM–100 μM), then incubated for a further 72 h. Following incubation 50 μL aliquots of medium were removed to a fresh 96-well plate. Cytotoxicity was determined using and LDH assay kit obtained from Promega, following the manufacturer's instructions for use. A 50 μL per well LDH substrate mixture was added and the plate left in darkness at room temperature for equilibration. Stop solution (50 μL) was added to all wells before reading the absorbance at 490 nm. A control of 100% lysis was determined for a set of untreated cells which were lysed through the addition of 20 μL lysis solution to the media 45 min prior to harvesting. Data was presented following calculation, as percentage cell lysis versus concentration of subject compound.

Estrogen receptor binding assay. ER α and ER β fluorescence polarization based competitor assay kits were obtained from Panvera at Invitrogen Life Technologies. The recombinant ER (insect expressed, full length, untagged human ER obtained from recombinant baculovirus-infected insect cells) and the fluorescent estrogen ligand were removed from the -80°C freezer and thawed on ice for 1 h prior to use. The fluorescent estrogen ligand (2 nM) was added to the ER (30 nM for ER α and 20 nM for ER β) and screening buffer (100 mM potassium phosphate (pH 7.4), 100 $\mu\text{g}/\text{mL}$ BGG, 0.02% NaN_3) was added to make up to a final volume that was dependant on the number of tubes used (number of tubes (e.g. 50) \times volume of complex in each tube (50 μL) = total volume (e.g. 2500 μL). Test compound (1 μL , concentration range 1 nM) to 100 μM) was added to 49 μL screening buffer in each borosilicate tube (6 mm diameter). To this 50 μL of the fluorescent estrogen/ER complex was added to make up a final volume of 100 μL and final concentration range for the test compound of 0.01 nM to 1 μM . A vehicle control contained 1% (v/v) of ethanol and a negative control contained 50 μL screening buffer and 50 μL fluorescent estrogen/ER complex. The negative control was used to determine the polarisation value when no competitor was present (theoretical

maximum polarization). The tubes were incubated in the dark at room temperature for 2 hours and were mixed by shaking on a plate shaker. The polarization values were read on a Beacon single-tube fluorescent polarization instrument with 485 nm excitation and 530 nm emission interference filters. For ER α and ER β , graphs of anisotropy (mA) versus competitor concentration were obtained for determination of IC₅₀ values.

Estrogenic activity: alkaline phosphatase assay. Following the procedure of Littlefield et al [24], human Ishikawa cells were maintained in Eagle's Minimum Essential Medium (MEM containing 10% vol/vol fetal bovine serum (FBS) and supplemented with 100 U/mL penicillin and 10 μ g/mL streptomycin, 2 mM glutamine and 1 mM sodium pyruvate. Twenty four hours before the start of the experiment, near confluent cells were changed to an estrogen-free medium (EFBM), A 1:1 mixture of phenol-free Ham's F-12 and Dulbecco's Modified Eagles Medium, together with the supplements listed above, and 5% calf serum, stripped of endogenous estrogens with dextran coated charcoal. On the day of the experiment, cells were harvested with 0.25% trypsin and plated in 96-well flat bottomed microtitre plates in EFBM at a density of 2.5×10^4 cells/well. Test compounds were dissolved in ethanol at 10^{-3} M, diluted with EFBM (final concentration of ethanol 0.1%) and filter sterilised. After addition of the test compounds, (plated in 50 μ L, added estradiol in 50 μ L, and blank medium to give a final volume 150 μ L) the cells were incubated at 37°C in a humidified atmosphere containing 95% O₂/5% CO₂ for 72 h. All experimental values were obtained in triplicate. The microtitre plates were then inverted and the growth medium removed. The plates were then rinsed by gentle immersion and swirling in 2 L of PBS (0.15 M NaCl, 10 mM sodium phosphate, pH 7.4). The plates were removed from the container, the residual saline in the plate was not removed, and the wash was repeated. The buffered saline was then shaken out, and the plate blotted on paper towel. The covers were replaced and the plates were placed at -80 C for at least 15 min. then thawed at room temperature for 5–10 min. The plates were then placed on ice and 50 μ L ice cold solution containing 50 mM *p*-nitrophenyl phosphate, 0.24 mM MgCl₂ and 1 M diethanolamine (pH 9.8) was added. The plates were warmed to room temperature (time zero), and the yellow colour from the production of *p*-nitrophenol was allowed to develop. The plates were monitored at 405 nm until maximum stimulation of the cells showed an absorbance of approximately 1.2.

Computational procedure

Ligand preparation: Compound **13c** was drawn using ACD/Chemsketch v10 [25] and converted to a SMILES string. A single conformer was generated with a final MMFF optimisation step for refinement of the compound, using Omega v2.1 [26].

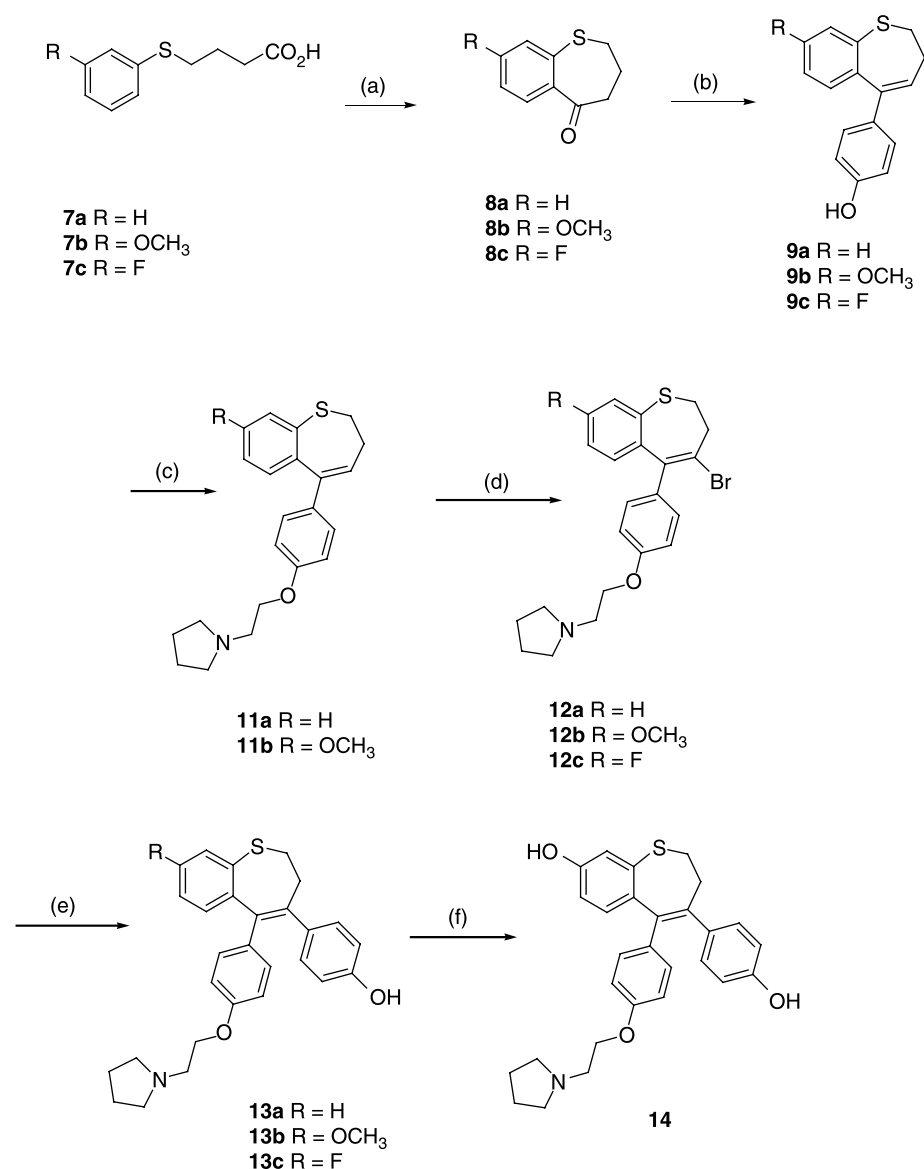
Receptor preparation: PDB entries 3ERT and 1QKN were downloaded from the Protein Data Bank (PDB). A single bridging water molecule held between Glu353 (Glu305) and Arg394 (Arg346) was retained in both isoforms. Addition and optimisation of hydrogen positions was carried out using MOE 2005.06 [27] ensuring all other atom positions remained fixed.

Docking: FRED v2.11 [28] was utilized in this study to dock the conformer in both estrogen receptor isoforms. All default values were applied with rigid-body optimisation of each ligand pose with Chemgauss2. The docked complexes for both isoforms were imported to Sybyl v6.91 [29] and a treated to a flexible ligand and active site docking run. The active site residues assessed previously using MOE 2005.06 were allowed to be flexible during the docking calculation. All other values were kept as default except the number of iterations, which was increased to 10,000. Finally the docked complex was optimised under the MMFF force field using Szybki v1.1 [30], with optimisation of free rotor torsions and of polar hydrogen positions close to the ligand.

Results and discussion

Chemistry

The synthesis of the benzothiepin type compounds is illustrated in Scheme 1. The phenylsulfanylbutyric acids **7a–c** (obtained by alkylation of the corresponding thiols with ethyl 4-bromobutyrate) were cyclised with polyphosphoric acid to afford the benzothiepin-5-ones **8a–c** respectively. Treatment of **8a–c** sequentially with *n*-butyllithium and the THP protected 4-bromophenol followed by acid workup afforded the phenolic products **9a–c** in good yield. Alkylation of the phenols **9a** and **9b** with 1-(2-chloroethyl)pyrrolidine, followed by bromination of the alkenes **11a** and **11b** with pyridinium hydrobromide perbromide afforded the vinyl bromide products **12a** and **12b** respectively. Subsequent Suzuki arylation of these bromides **12a–b** with 4-hydroxyphenyl boronic acid resulted in isolation of the required benzothiepin products **13a** and **13b** in good yield (58–72%), (Table I). The fluoro substituted product **12c** required initial vinyl bromination of the phenol **9c** and subsequent alkylation of the resulting bromide **10** to afford the required product **12c**, (Scheme 2).

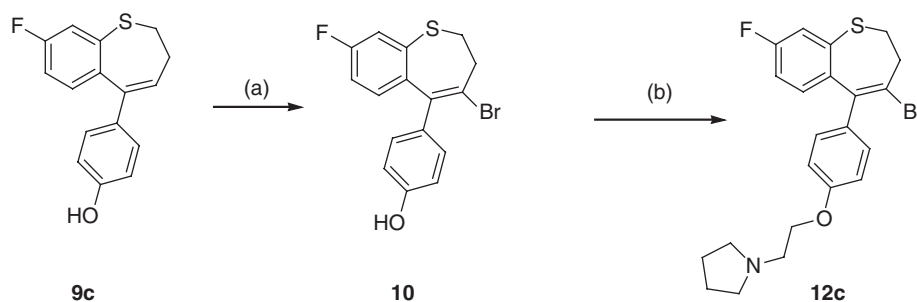


Scheme 1. Synthesis of Benzothiepins **13a–c** and **14**. Scheme reagents: (a) R = H, F; PPA, 110°C, 4h; R = OMe, Eaton's reagent, 80°C, 2h; (b) i) *n*BuLi, 4-Br-C₆H₄-OTHP, THF; ii) HCl, MeOH. (c) K₂CO₃, 1-(2-chloroethyl)pyrrolidine hydrochloride, acetone, 24h, 85°C; (d) PyHBr₃, CH₂Cl₂, 20°, 18h; (e) Pd(PPh₃)₃, 4-HOC₆H₄B(OH)₂, Na₂CO₃(aq), THF; (f) BBr₃, CH₂Cl₂.

Table I. Antiproliferative effects and estrogen receptor binding affinities for benzothiepin type SERMs.

Compound	Yield (%)	MCF-7 IC ₅₀ (μM) ^a	ERα/IC ₅₀ (nM) ^b	ERβ/IC ₅₀ (nM) ^b	ER β/α
6	52	1.584 ± 0.111	8.77	46.1	0.19
13a	72	0.0698 ± 0.006	8.4	3.8	2.21
13b	58	0.210 ± 0.167	9.1	2.4	3.79
13c	40	0.0185 ± 0.001	6.76	0.7	9.65
14	62	0.306 ± 0.003	10.0	6.5	1.54
Tamoxifen^c	–	4.12 ± 0.038	70	170	0.41

^aExperimental values represent the average for two experiments performed in triplicate along with the standard deviation (SD) between the assay values. Values without SD values were run once. IC₅₀ values are half maximal inhibitory concentrations required to block the growth stimulation of MCF-7 cells; ^b Values are an average of at least nine replicate experiments for ERα with typical standard errors below 15%; and six replicate experiments for ERβ with typical standard errors below 15%; ^cThe value for tamoxifen IC₅₀ 4.12 ± 0.38 μM is in good agreement with the reported IC₅₀ value for tamoxifen using the MTT assay on human MCF-7 cells.



Scheme 2. Synthesis of Benzothiepin **12c**. Scheme reagents: (a) PyHBr_3 , CH_2Cl_2 , 20° , 18h; (b) K_2CO_3 , 1-(2-chloroethyl)pyrrolidine hydrochloride, acetone, 24h, 85°C .

The diphenolic compound **14** was obtained in 62% yield by demethylation of **13b** with boron tribromide in dichloromethane. This reaction required careful tlc monitoring as some C7- brominated product was also formed in the reaction. Alternative demethylation reagents such as pyridine hydrochloride and boron trifluoride dimethylsulfide were less successful in this reaction.

Biochemistry

The compounds were evaluated in a series of estrogen dependent *in vitro* assays which measured their affinity for the estrogen receptors and also their ability to act as functional antagonists or agonists of estrogen.

Antiproliferative activity in MCF-7 breast cancer cells.

Compounds **13a–c**, **14** containing the benzothiepin-derived molecular scaffold were initially assessed for their antiproliferative action using the ER expressing (ER dependent) human MCF-7 breast cancer cell line and the results are displayed in Table I. These compounds are considerably more potent than tamoxifen, ($\text{IC}_{50} = 4.120 \mu\text{M}$) and the initial benzoxepin compound **6** ($\text{IC}_{50} = 1.584 \mu\text{M}$). The 8-fluoro substituted compound **13c** proved to be the most active compound in the series with $\text{IC}_{50} = 18.5 \text{ nM}$. The fluoro substituent at C-8 would prevent metabolic hydroxylation occurring at this position. The 8-unsubstituted compound **13a** also displayed potent activity with $\text{IC}_{50} = 69 \text{ nM}$. Compounds **13b** (methoxy substituted) and **14** (containing phenolic substituents on both Rings B and C) were moderately active ($\text{IC}_{50} = 209$ and 306 nM respectively). The replacement of the OH group on compound **14** ($\text{cLog P} = 5.99$) with lipophilic fluorine as in **13c** ($\text{cLog P} = 6.86$) results in a decrease in the IC_{50} value from 306 nM to 18.5 nM indicating the positive effect of the fluorine on the antiproliferative activity. These results compare very favourably with the antiproliferative activity reported for tamoxifen [31], fluorotamoxifen[32,33], benzocycloheptenes and related dibenzothiepins. [34–36].

In order to determine whether the antiproliferative effects elicited by the benzothiepin compounds was

mediated through the ER, the most potent examples (compounds **13c** and **14**) were tested in hormone independent MDA-MB 231 breast cancer cells and displayed IC_{50} values of 120 nM and $1.47 \mu\text{M}$ respectively. The value obtained for compound **13c** in this assay suggests that that compound may function through alternative antiproliferative mechanisms, independent of the ER. The cytotoxicity of the compounds was also determined in the standard LDH assay as previously reported as we wished to confirm that the antiproliferative effects of the compounds were due to cytostasis rather than cellular necrosis. The compounds all demonstrated low cytotoxicity profiles suggesting that their action is cytostatic rather than cytotoxic, (Table I). Typical values for cytotoxic induced antiproliferative effects for compound **13a** at $10 \mu\text{M}$ is 8%, whereas value for tamoxifen is determined to be 24% at testing concentration of $10 \mu\text{M}$. The cytotoxicity values obtained were less than that for tamoxifen for all compounds except for compound **13c** (31% at $10 \mu\text{M}$) which is one of the most potent antiproliferative compounds in the series evaluated. The corresponding values in the MDA-MB 231 assay for compounds **13c** and **14** were 17% and 11% respectively again indicating a relatively low level of cytotoxicity for these benzothiepins.

Estrogen receptor binding studies. Estrogen receptor binding studies were carried out with $\text{ER}\alpha$ and $\text{ER}\beta$ using a fluorescence polarization procedure[37]. The displacement of fluorescein labeled estradiol (fluoromone) in a competitive binding assay from the human recombinant full length receptor proteins $\text{ER}\alpha$ and $\text{ER}\beta$ expressed from baculovirus – infected insect cells by the synthesized ligands was observed as a decrease in polarization values. Compounds **13a–c** and **14** containing the benzothiepin-derived molecular scaffolds were effective $\text{ER}\alpha$ ligands with IC_{50} values in the range $8–10 \text{ nM}$ (Table I), indicating that the benzothiepin sulfur is equally as effective as the benzoxepin oxygen in facilitating binding of the ligands to the $\text{ER}\alpha$ as also reported for benzoxathiins[17]. The $\text{ER}\alpha$ and $\text{ER}\beta$ binding values for compounds **13a**, **13b** and **14** were also confirmed in a competitive radiometric

Table II. Summary of key Ligand-Protein contacts^a.

Compound	Isoform	Asp 351 (Ala 257)	Glu 353 (Glu 260)	Arg 394 (Arg 301)	His 524 (His 430)
13c	α	4.7	2.8	2.5	3.0
13c	β	3.7	2.3	5.1	3.2

^aResidues depicted are those present in crystal structure 3ERT, except those in brackets denoting residues of crystal structure 1QKN. Bold font indicates H-bonding distances.

binding assay. The introduction of the more lipophilic fluorine at C-8 in compound **13c** (cLog P = 6.86) resulted in considerably more potent activity in the MTT assay and ER α and ER β binding assays than compound **14** (cLog P = 6.99). The benzothiepin-derived compounds were all potent ligands for ER β with IC₅₀ values in the range 0.7–6.5 nM. The twelve-fold selectivity of compound **13c** for ER β was attributed to the presence of the F substituent at position 8 in Ring B of the rigid structure (as discussed in the molecular modeling and docking section below) and compare very favourably with the ER binding data reported for fluorotamoxifen (IC₅₀ 5 μ M)[32]. The corresponding diphenolic compound **14** showed much lower selectivity with β/α ratio of 1.5 respectively which is closer to the β/α ratio value of 1.67 obtained for 4-hydroxytamoxifen in this experiment.

Estrogenic stimulation. The estrogen stimulation and antagonistic properties of the most active compounds **13c** and **14** were determined in the estrogen bioassay which is based on the stimulation of alkaline phosphatase (AlkP) in the Ishikawa human endometrial adenocarcinoma cell line [24]. Compounds are tested as estrogen antagonists by their effect on the

inhibition of estradiol stimulation in the Ishikawa cells in a dose-dependent manner. Compounds **13c** and **14** were much more potent than tamoxifen as estrogen antagonists with IC₅₀ values of 7.4 nM and 6.4 nM respectively and correlates with the potent ER α and ER β binding observed for these compounds. The estrogenic stimulatory properties of these compounds could be determined in the Ishikawa cells by measuring the stimulation of alkaline phosphatase (AlkP) in these cells in the absence of estradiol. Compounds **13c** and **14** displayed minimal stimulatory values of 0.2% and 3.3% respectively. The results from this AlkP assay for estrogen antagonism and stimulation provide useful information in selection of the structural features for optimum antiestrogenic activity without associated adverse estrogenic effect on tissues such as the uterus.

Molecular modelling

To rationalize the observed biological activity of the most active compound **13c**, a flexible computational investigation was undertaken. Currently there are no crystal structures of human ER α/β with the same antagonist co-crystallized and thus to compare docking of the same ligand in different isoforms remains difficult. To overcome this problem receptor flexibility

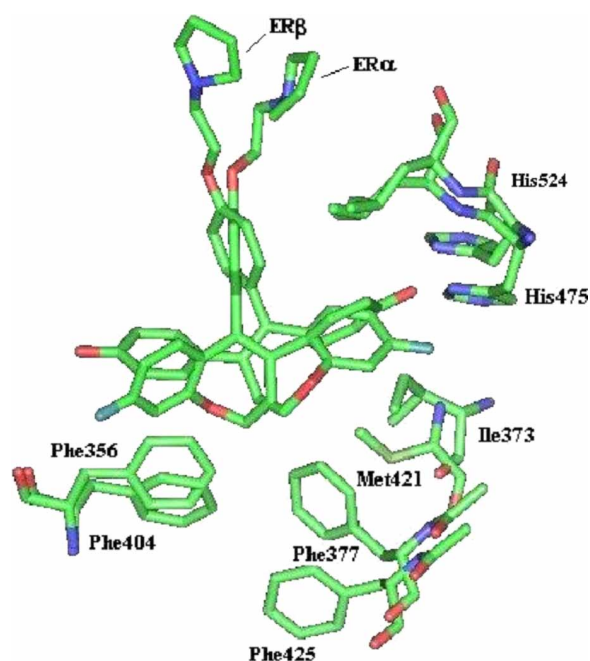


Figure 2. Docked complex of compound **13c** in active site of ER α and ER β superposed.

must be accounted for in the docking process. Utilising Flexidock (Sybyl6.91), both the ligand and active site residues were allowed to move flexibly during docking. Finally the resulting pose was optimised using Szybki v1.1. The resulting pose was selected and atomic interactions were analysed by Ligand Protein Contacts (LPC) software [38]. The residues depicted are those that are known to be key in the binding process for both agonists/antagonists: Asp351, Glu353 and Arg394 (anchor the ligand in the active site), His524 (additionally important in ligand binding process). Table II illustrates the interactions made by each ligand with both receptor isoforms.

As illustrated in Figure 2, compound **13c** docks in a typical antiestrogenic manner when compared with 4-hydroxytamoxifen or raloxifene. What is immediately apparent is the inversion or flipping of the rings A and C placing the fluoro and hydroxy substituents at opposite ends. Mewshaw *et al* [39] have observed this type of 'ring-flipping' previously due to the pseudosymmetry of the 2-phenylanthralene scaffold. Consequently, selectivity for ER β was also observed with the series due to this A & C 'ring-flipping'. In the case of the benzothiepin scaffold, the ligand docks in ER α forming a hydrogen bond with His524 whereas in ER β none is observed. Instead, the hydroxy group of ring C forms a strong hydrogen bonding interaction with Ala350 and Glu353. This orientation within ER α reduces bonding with Asp351 compared with in ER β as illustrated in Table II.

The substitution of Met421 (ER α) with Ile373 (ER β) causes steric interaction with the ligand, supplementary to that imposed by Phe356 and Phe377, preventing the ligand from docking in the same orientation as observed with ER α . It has been suggested previously by a number of groups [40,41] that enhanced ER β selectivity is achieved through a ligand that can differentiate between ER α Met421 and ER β Ile373. Compound **13c** exhibits these preferences and thus was observed to be ~ 10-fold selective for ER β .

Conclusion

In summary, this study demonstrates that benzothiepin derived molecular scaffolds can provide useful estrogen receptor modulators. The compounds exhibited potent activity in an antiproliferative assay against the MCF-7 breast cancer cell line and specificity in binding to the ER β subtype. The potent β -specific binding of compound **13c** is rationalized in a docking study which demonstrated the effect of the fluorine substituent in differentiation of ligand binding to the ER receptor subtypes. These results should facilitate the rational design of further ER β subtype specific ligands.

Acknowledgements

We are very grateful to Professor Richard Hochberg at Yale University Medical School, for kindly facilitating

the alkaline phosphatase experiments and for the generous gift of the Ishikawa cells. This work was supported through funding from the Trinity College IITAC research initiative (HEA PRTLII), with additional support for computational facilities from the Wellcome Trust.

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